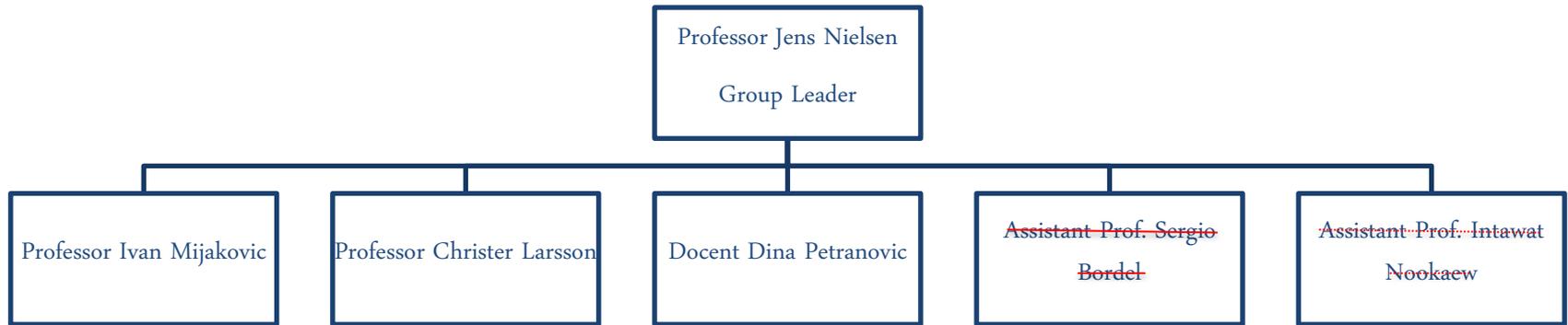


Nielsen Lab

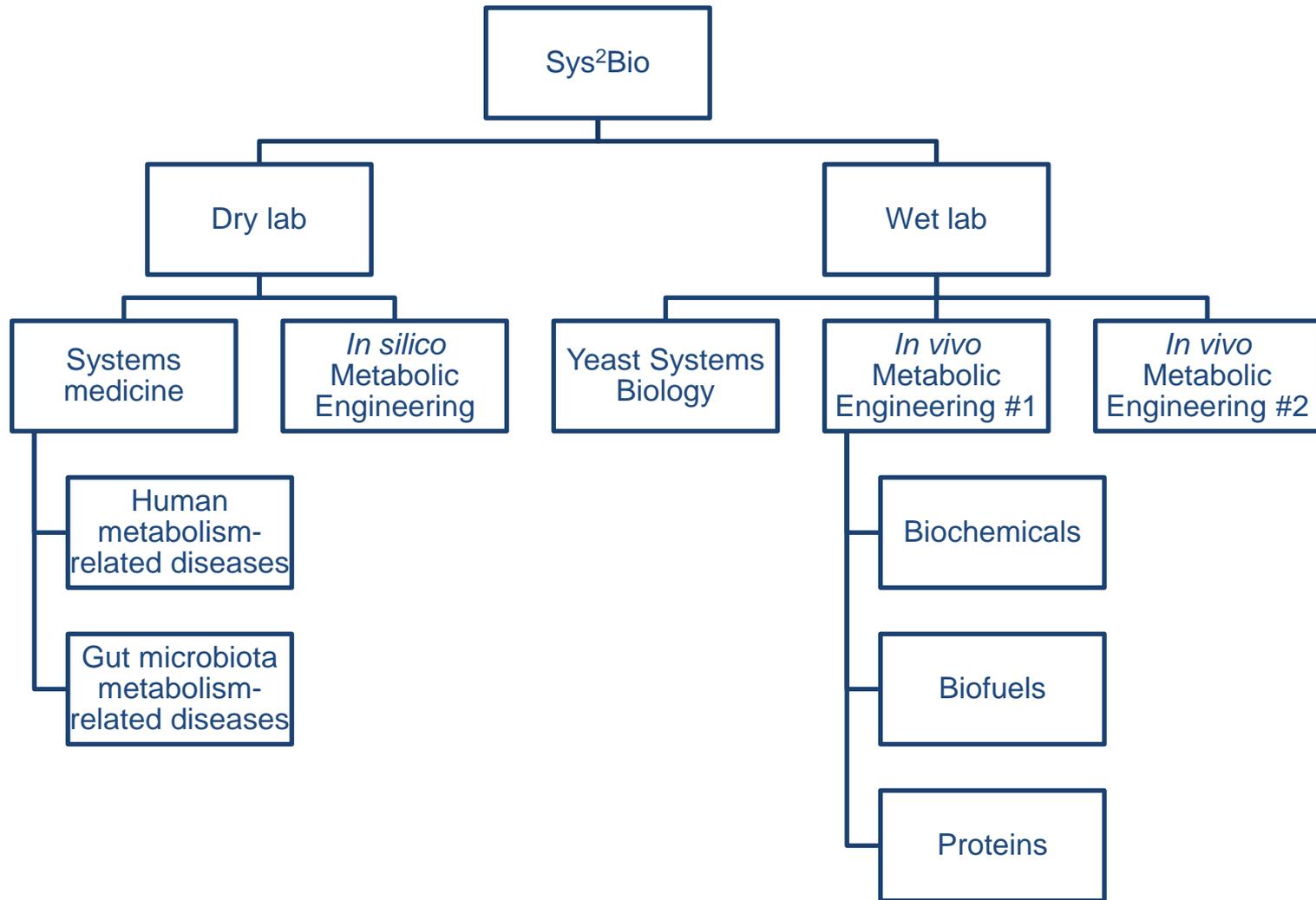
Systems and Synthetic Biology
Chalmers University of Technology
Sweden

Systems and Synthetic Biology

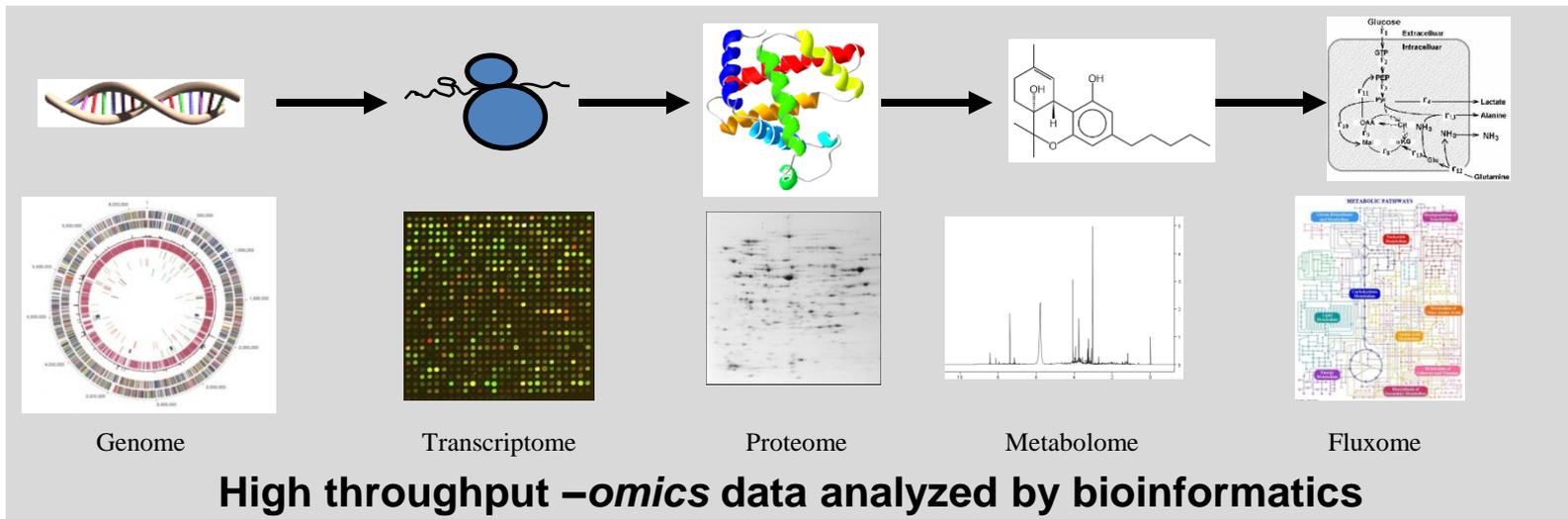


19 Post Docs
 25 PhD students
 5 Visiting PhD students
 8 MSc students
 7 Research Engineers/Technicians
 2 Admin

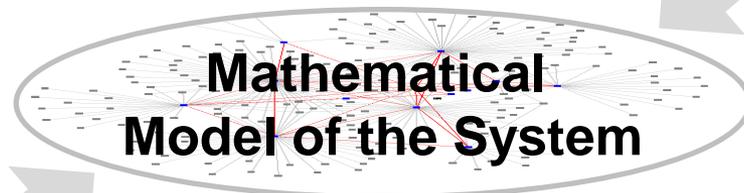
Nielsen Lab Organization



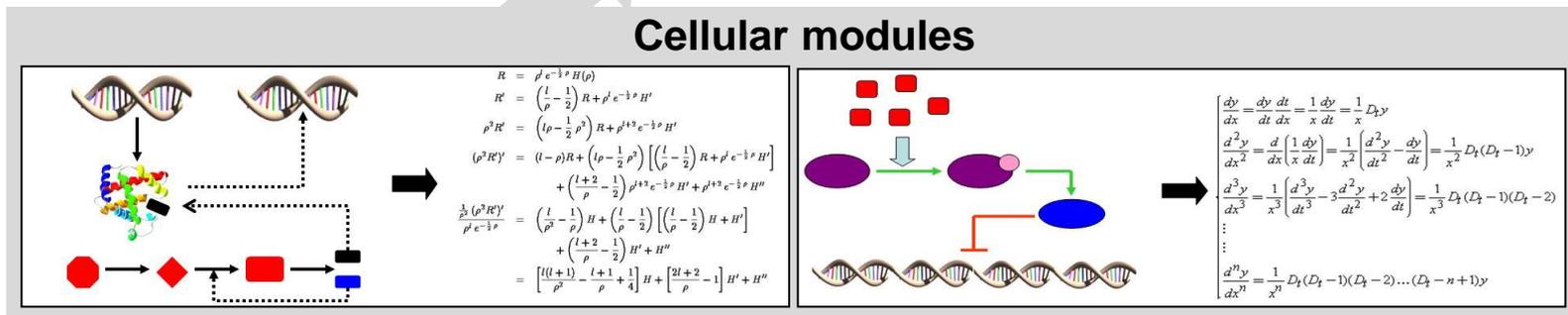
Top-Down & Bottom-Up SB



Discovery Based
Top-down Approach

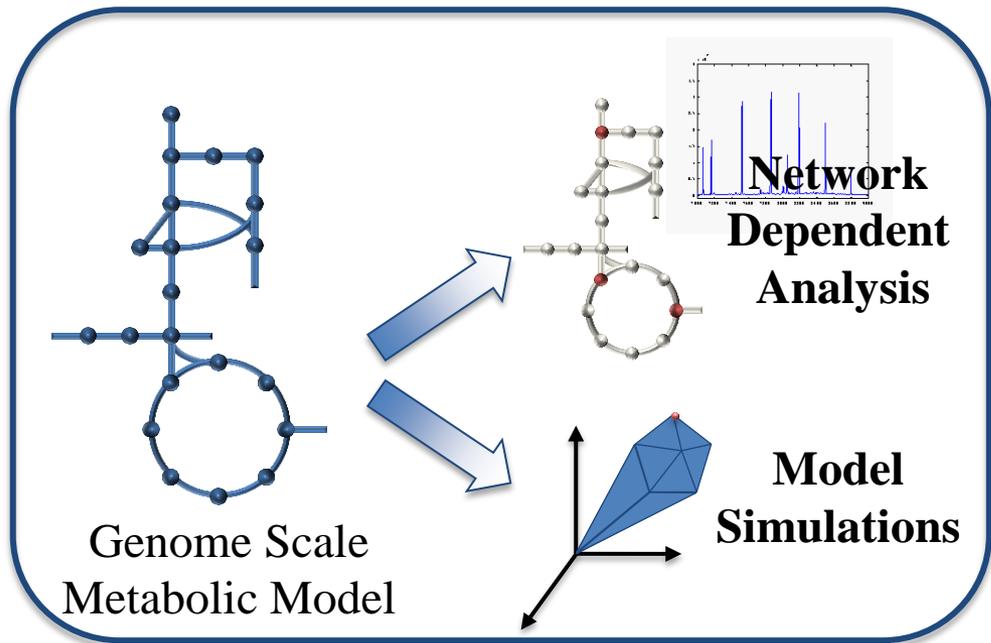
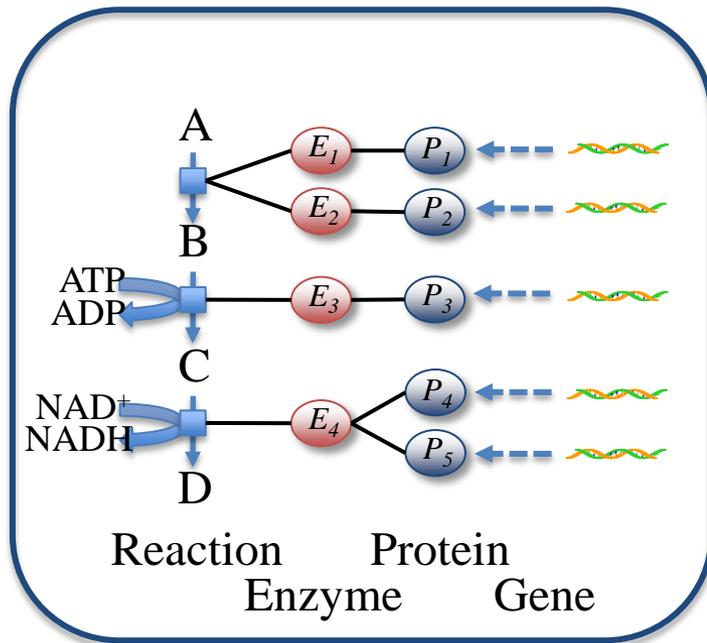
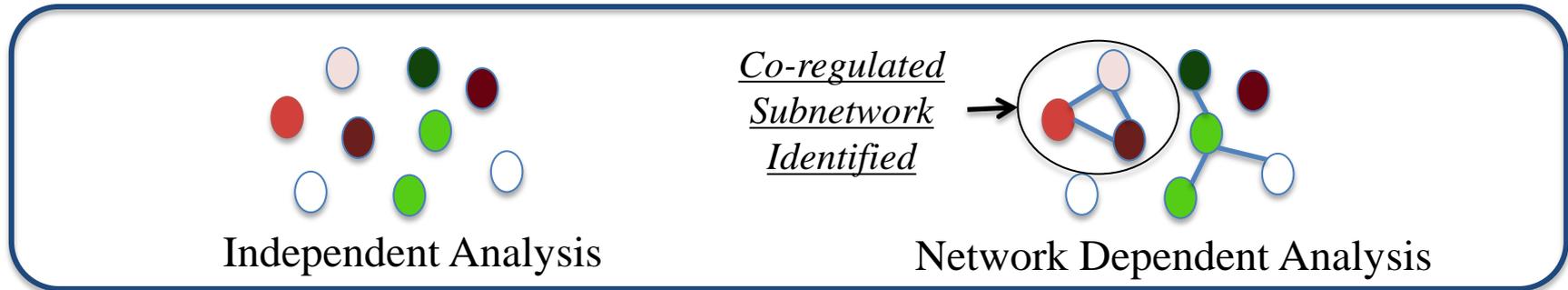


Hypothesis Based
Bottom-up Approach



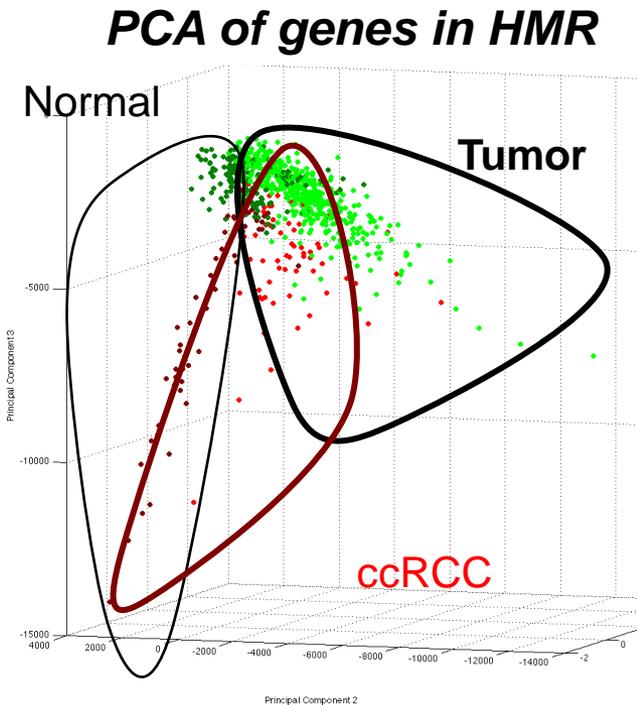
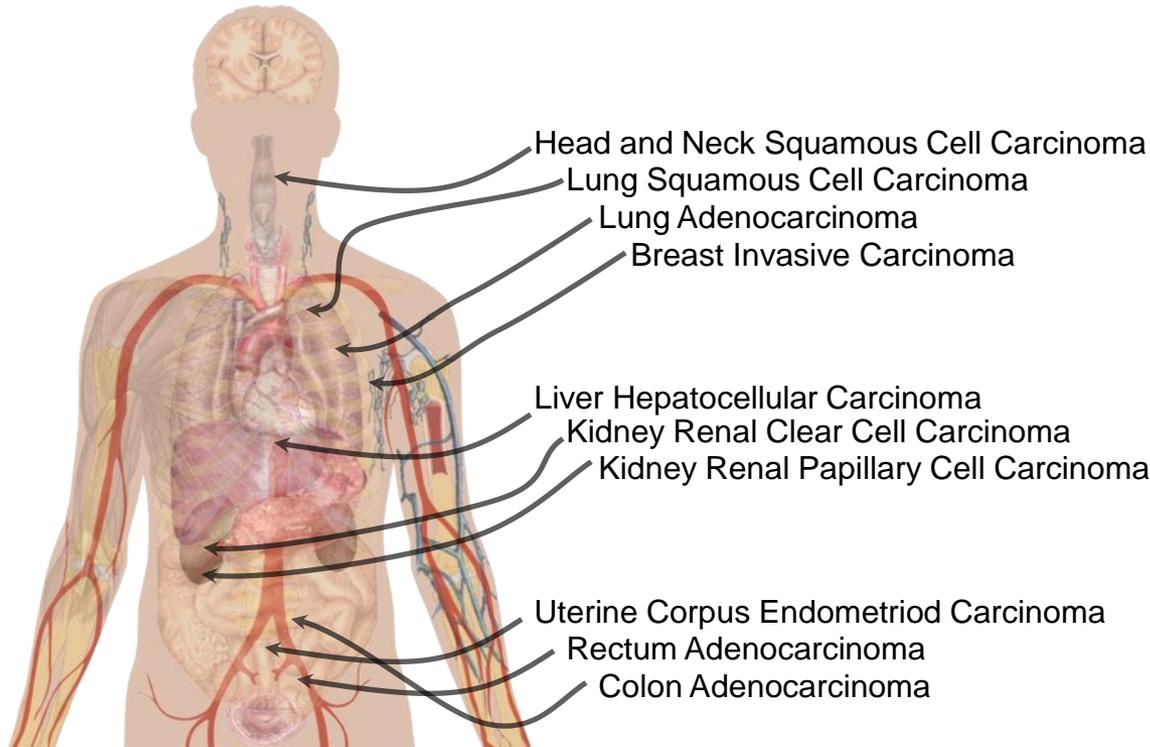
Genome-Scale Metabolic Models

Genome-scale metabolic models (GEMs) provide gene-protein-reaction connections and hereby allow for context dependent analysis



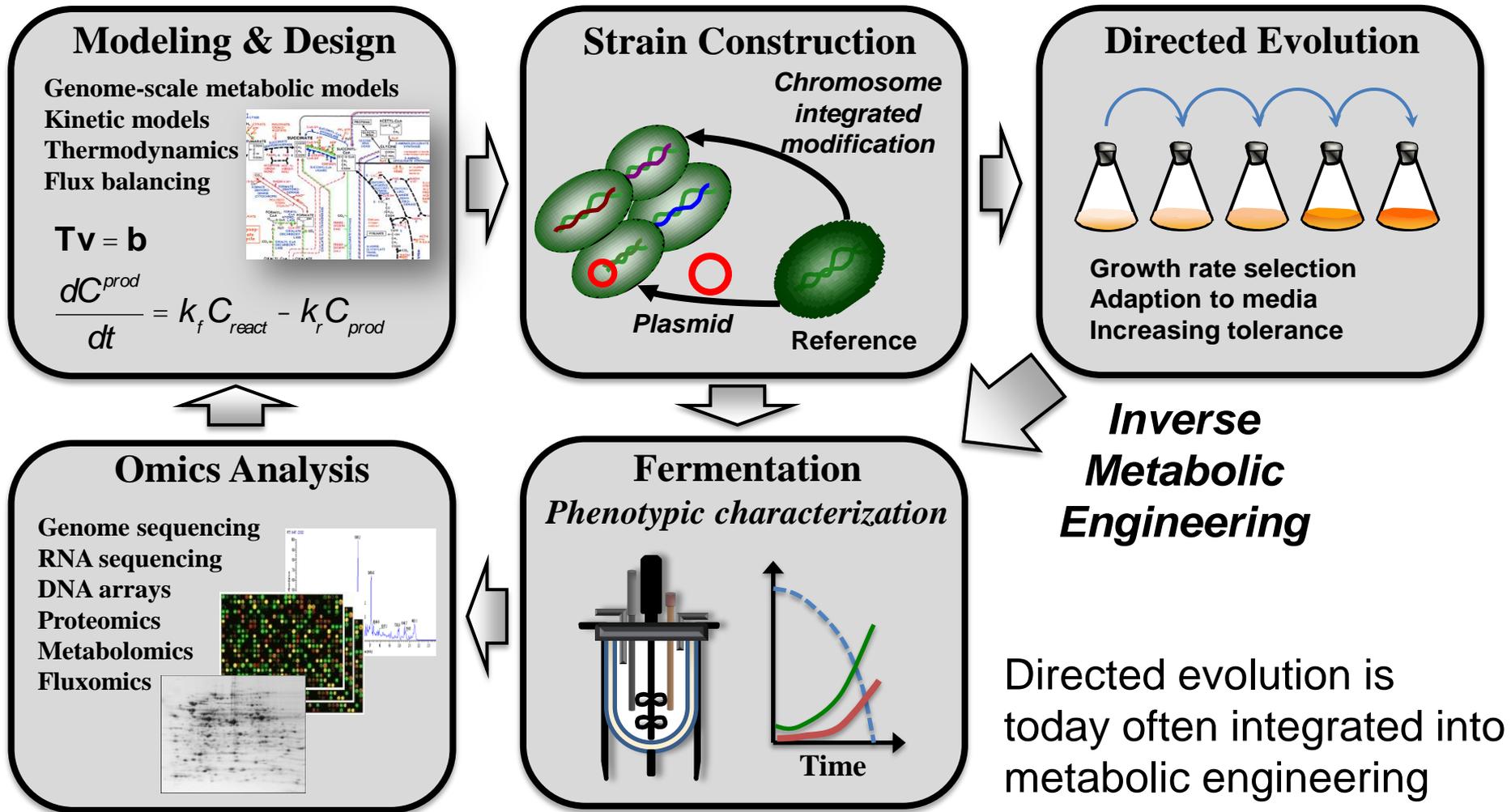
Is Metabolism of Cancer Cells Different from Normal Cells

RNAseq “*metabolic*” transcriptome of 10 cancers and control



Clear cell Renal Cell Carcinoma (kidney cancer) deviates and do not show a clear distinction in terms of transcription of metabolic genes

Metabolic Engineering

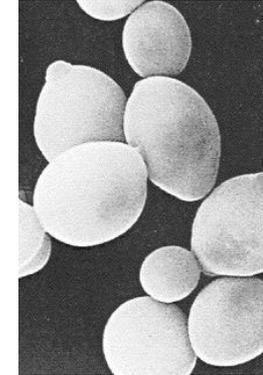


Directed evolution is today often integrated into metabolic engineering

Yeast as a Platform Cell Factory

Many advantages:

- Extremely well-characterized
- Many online databases with information on genome, as well as different omics data
- Genetically tractable
- GRAS (Generally Regarded as Safe)
- **Robust industrial organism**



A widely used cell factory

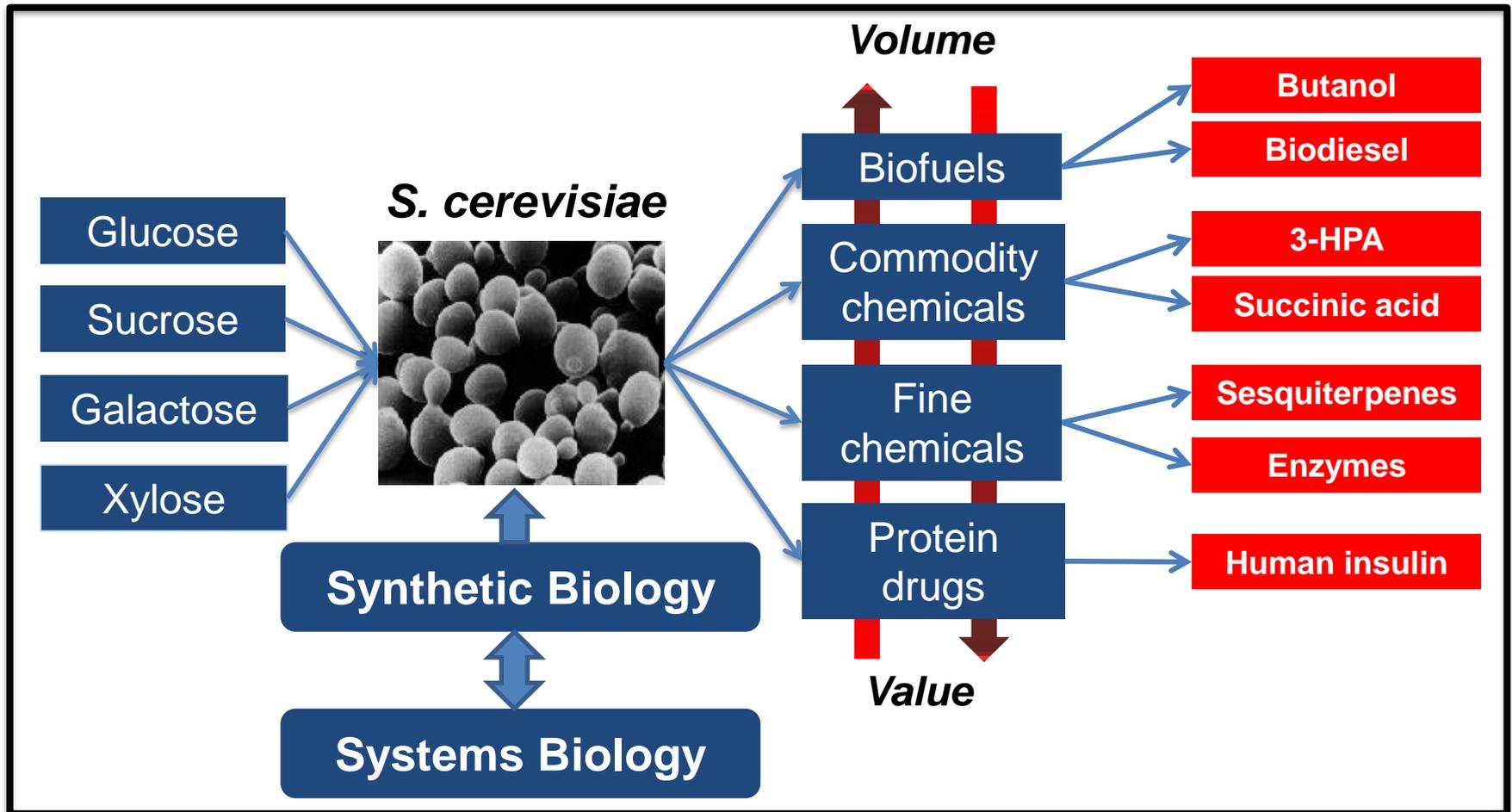
- ◆ Bioethanol
- ◆ Baker's yeast
- ◆ Wine & Beer
- ◆ Resveratrol
- ◆ Insulin precursors
- ◆ Vaccines (HPV, Hepatitis)

Ongoing developments of novel cell factories:

- ◆ **Fuels** (butanol, biodiesel)
- ◆ **Commodity chemicals** (malate, succinate, 3-OH propionic acid)
- ◆ **Fine chemicals** (isoprenoids)
- ◆ **Food ingredients** (PUFAs)
- ◆ **Protein drugs**

Metabolic Engineering of Yeast

Our objective is to establish an extensive technology base for wider use of yeast as platform cell factory and demonstrate its use for development of diverse products



Butanol production and tolerance in *Saccharomyces cerevisiae*

Payam Ghiaci

Francisca Lameiras

Joakim Norbeck

Intawat Nookaew

Christer Larsson

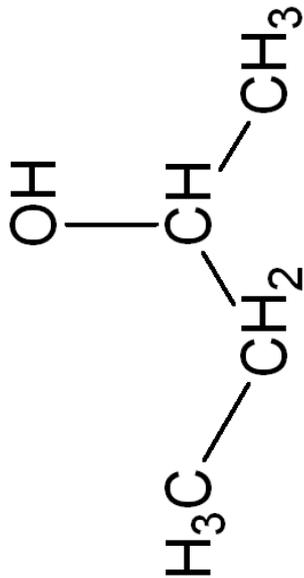
*Chalmers University of Technology, Dept Chemical and
Biological Engineering - Systems Biology, Gothenburg,
Sweden*

Bio-fuels; Butanol vs Ethanol



Why butanol?

- Butanol has a higher energy content compared to ethanol
- Lower water absorption and volatility compared to ethanol
- Existing distribution systems can be used
- Can be used in conventional engines without or with less modifications

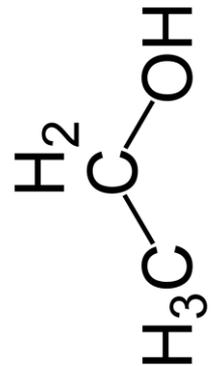


Why not butanol?

Butanol is very toxic to the producing organisms

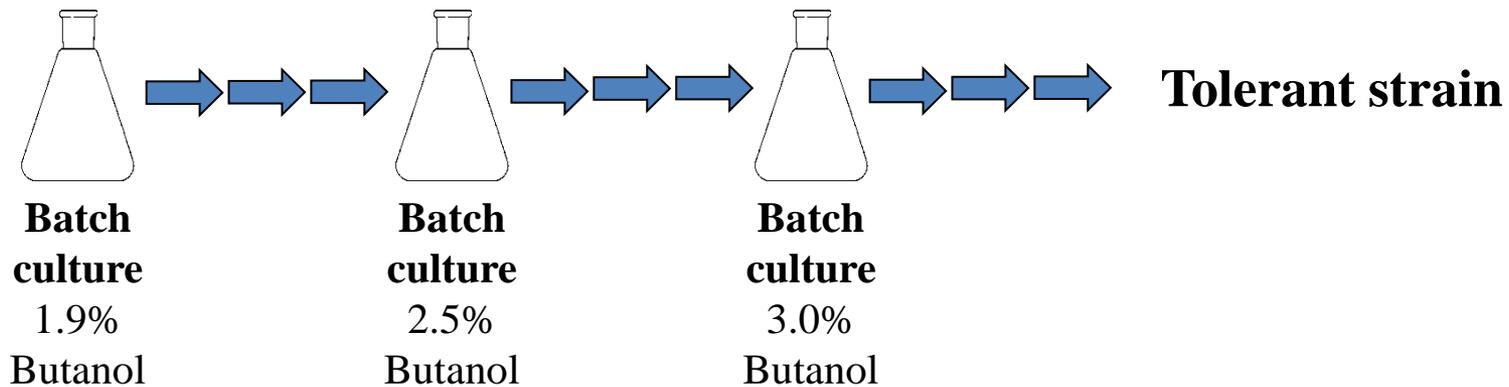
Why yeast?

- Outstanding history of human usage and exploitation under large-scale industrial conditions
- Can cope with harsh and/or nutrient poor conditions such as, *e.g.* Lignocellulosic substrates
- Amenable to genetic manipulations



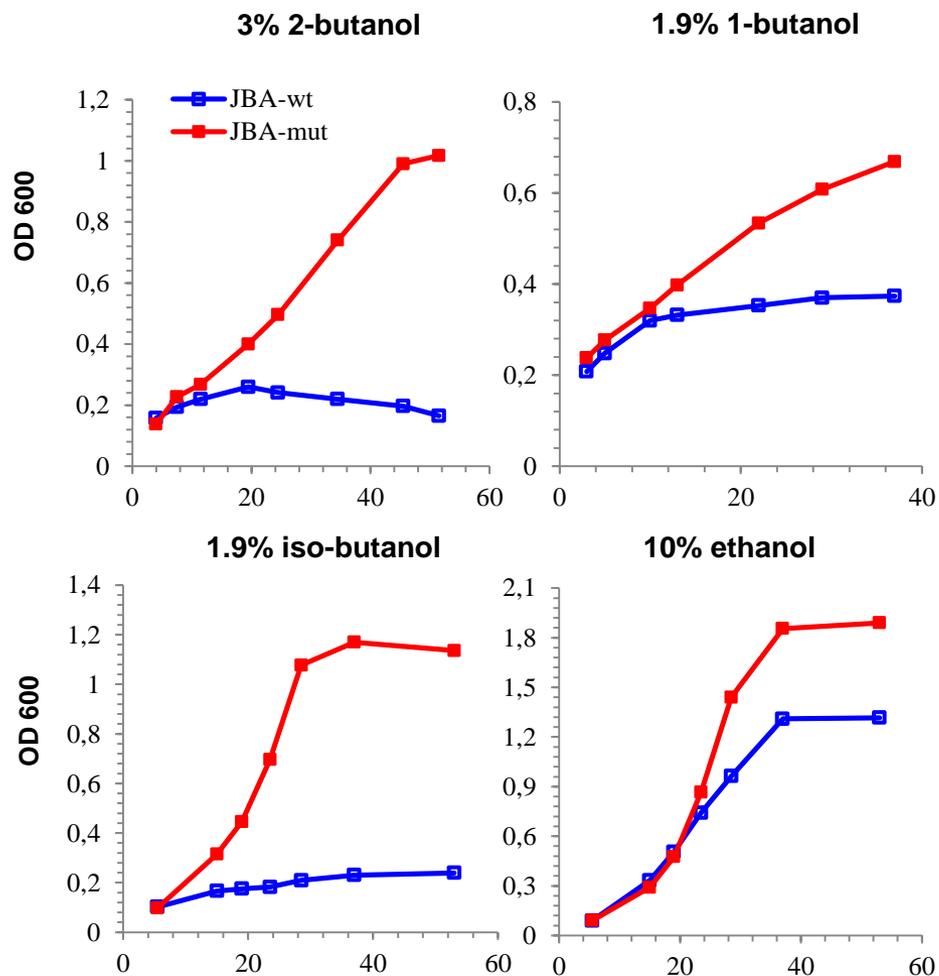
Part 1: Tolerant yeast

Evolving a 2-butanol tolerant yeast via evolutionary engineering



- 30 sequential batch cultures
- About 100 generations within 24 days
- Industrial baker's yeast strain

Comparison of growth in the presence of various alcohols between wild-type and the evolved 2-butanol tolerant strain



Characterization of evolved mutant; protein expression and lipid analysis.

Growth conditions

1. Chemostat culture, $D = 0.1 \text{ h}^{-1}$, with 2.5% 2-butanol
2. Batch culture with 1.2% 2-butanol (μ identical between mutant and wild-type)

Protein expression profile by MS; comparison wild-type vs evolved mutant

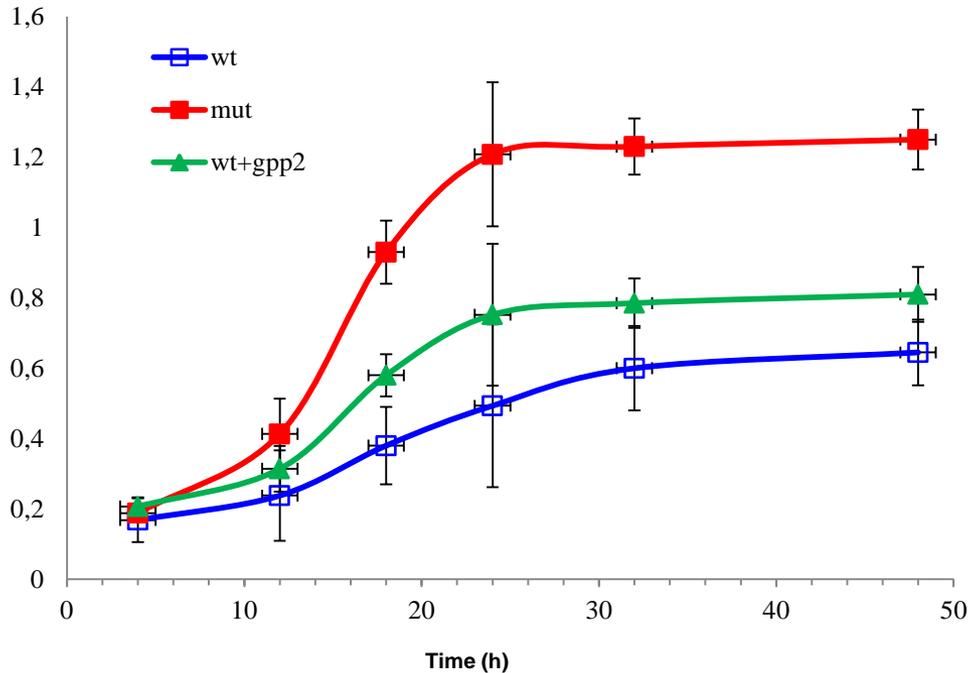
Gene	Corresponding protein	Fold change	Mitochondrial location
PIM1	Lon protease homolog	2.86	X
MAM33	Mitochondrial acidic protein MAM33	2.72	X
CYT1	Cytochrome c1, heme protein	2.09	X
GLO1	Lactoylglutathione lyase	2.01	
HSP42	Heat shock protein 42	1.97 *	
AIM2	Protein AIM2	1.94	X
FUN30	Uncharacterized ATP-dependent helicase FUN30	1.88 *	X
HOR2	(DL)-glycerol-3-phosphatase 2	1.81 *	
MCR1	NADH-cytochrome b5 reductase 2	1.80 *	X
GLK1	Glucokinase-1	1.74 *	
MRPL38	54S ribosomal protein L38	1.70 *	X
QCR6	Cytochrome b-c1 complex subunit 6	1.67	X
EDE1	EH domain-containing and endocytosis protein 1	1.65	
MSS116	ATP-dependent RNA helicase MSS116	1.65 *	X
YPL088W	Putative aryl-alcohol dehydrogenase YPL088W	1.64 *	
ATP4	ATP synthase subunit 4	1.62	X
ATP17	ATP synthase subunit f	1.62 *	X
PEP4	Saccharopepsin	1.62	
LSP1	Sphingolipid long chain base-responsive protein LSP1	1.62	
QCR2	Cytochrome b-c1 complex subunit 2	1.61	X
COX4	Cytochrome c oxidase subunit 4	1.59	X
ZWF1	Glucose-6-phosphate 1-dehydrogenase	1.59	
ECM33	Cell wall protein ECM33	1.58	
GVP36	Protein GVP36	1.57	
CCP1	Cytochrome c peroxidase	1.57	X
CAR2	Ornithine aminotransferase	1.57 *	
AAC2	ADP, ATP carrier protein 2	1.56	X
CYC1	Cytochrome c iso-1	1.56 *	X
ATP1	ATP synthase subunit alpha	1.55 *	X
ATP2	ATP synthase subunit beta	1.54 *	X
CPR3	Peptidyl-prolyl cis-trans isomerase C	1.54	X
KGD1	2-oxoglutarate dehydrogenase	1.54	X
QCR7	Cytochrome b-c1 complex subunit 7	1.53	X
MRP8	Uncharacterized protein MRP8	1.51 *	

- ~1100 proteins quantified (also membrane proteins)
- Expression of ribosomal proteins was not changed
- 34 proteins were up-regulated 1.5 fold or more ($\geq 95\%$ significance)
- 21 of these were mitochondrial and 12 components of respiratory chain
- Glo1, Hsp42 and Gpp2 were the most up-regulated non-mitochondrial proteins
- 12 proteins down-regulated 1.5 times or more ($\geq 95\%$ significance)

P. Ghiaci, J. Norbeck and C. Larsson. *Biotechnology for biofuels*, 2013, **6**:101

Verification of protein expression data by individual overexpression of Glo1, Hsp4, Gpp2 and Hap4

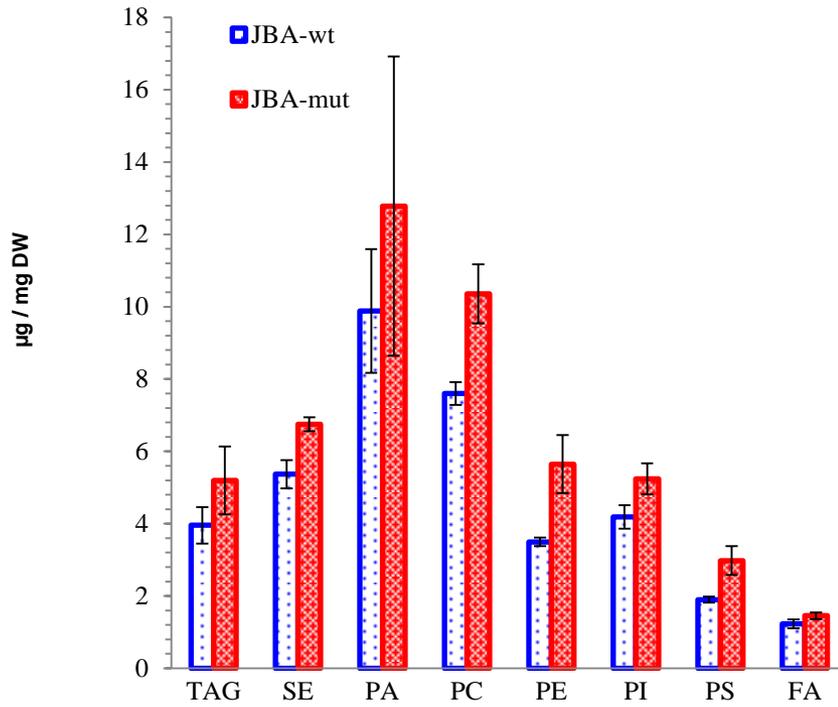
3% 2-butanol



Gpp2 overexpression improved 2-butanol tolerance

Glo1, Hsp42 and Hap4 overexpression did not improve 2-butanol tolerance

Lipid analysis; comparison wild-type vs evolved mutant



Tendency to a somewhat higher content of lipids in the tolerant mutant (not significant)

TAG = triacylglycerol

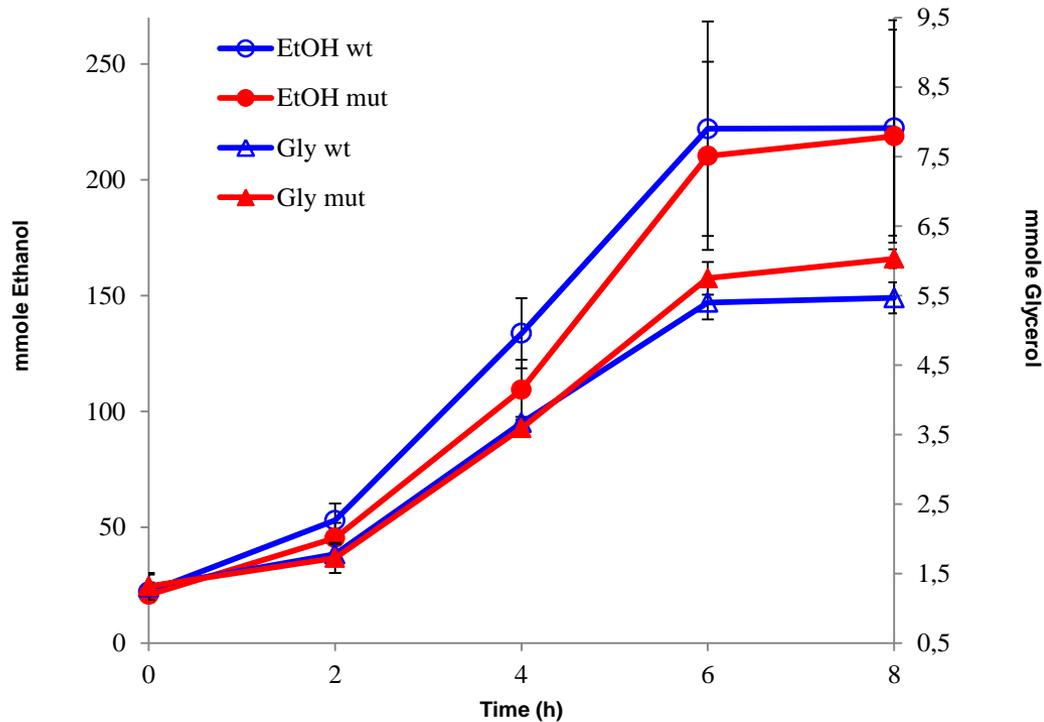
SE = steryl ester

PA, PC, PE, PI, PS = different phospholipids

FA = free fatty acid

ES = ergosterol

Growth characteristics; comparison wild-type vs evolved mutant during growth with 1.2% 2-butanol



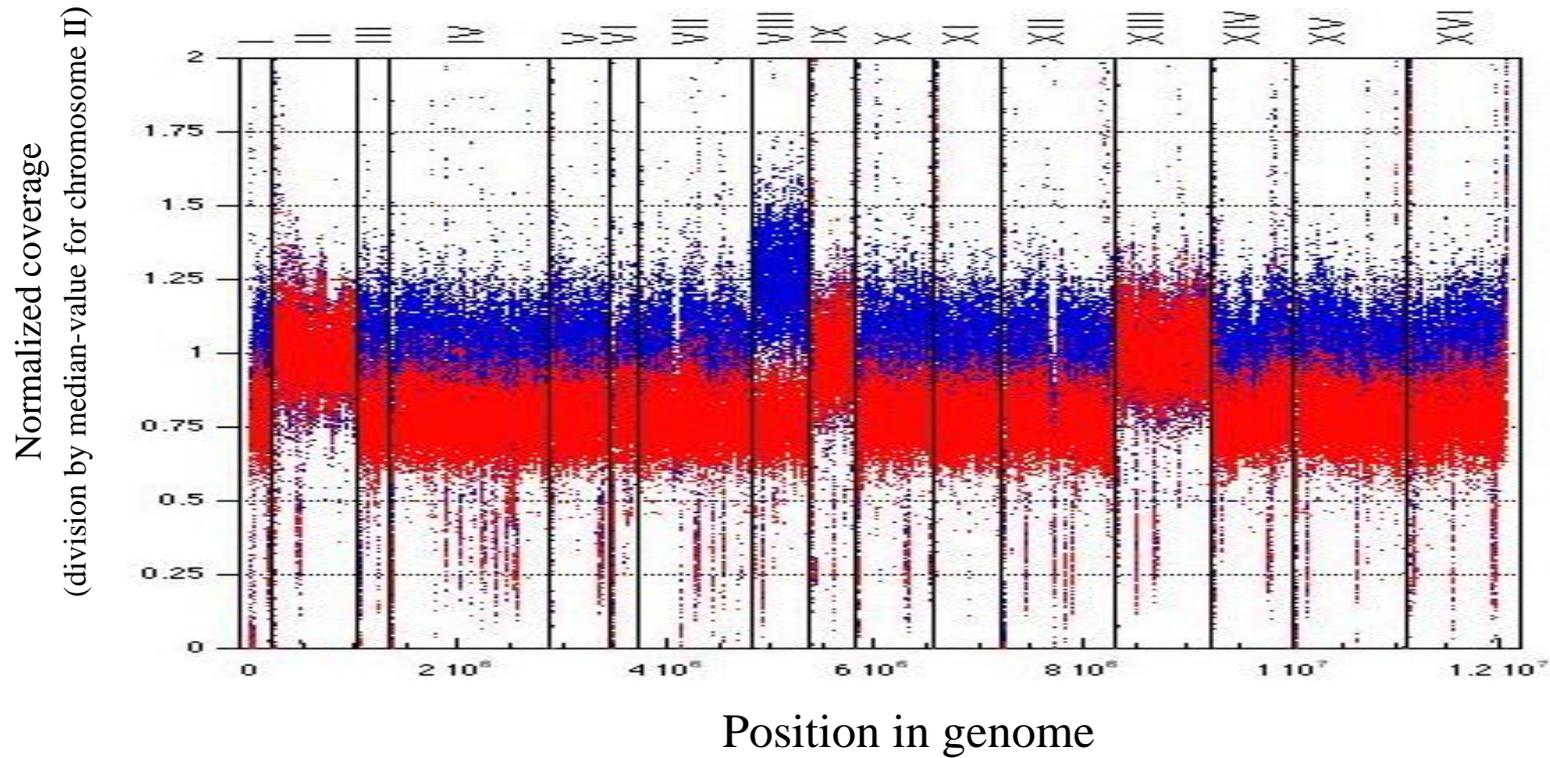
Growth characteristics were very similar

2-butanol tolerant mutant had a slightly lower ethanol and higher glycerol production consistent with proteomics data

Whole genome sequencing and comparison between wild-type vs evolved mutant

Butanol tolerance correlates with a loss of ploidy for most chromosomes.

(we have sequenced with approximately 1000 reads/bp)



Plot of normalized sequence coverage (100 bp average) against position in genome.

Blue is JBA-wt and red is JBA-mut.

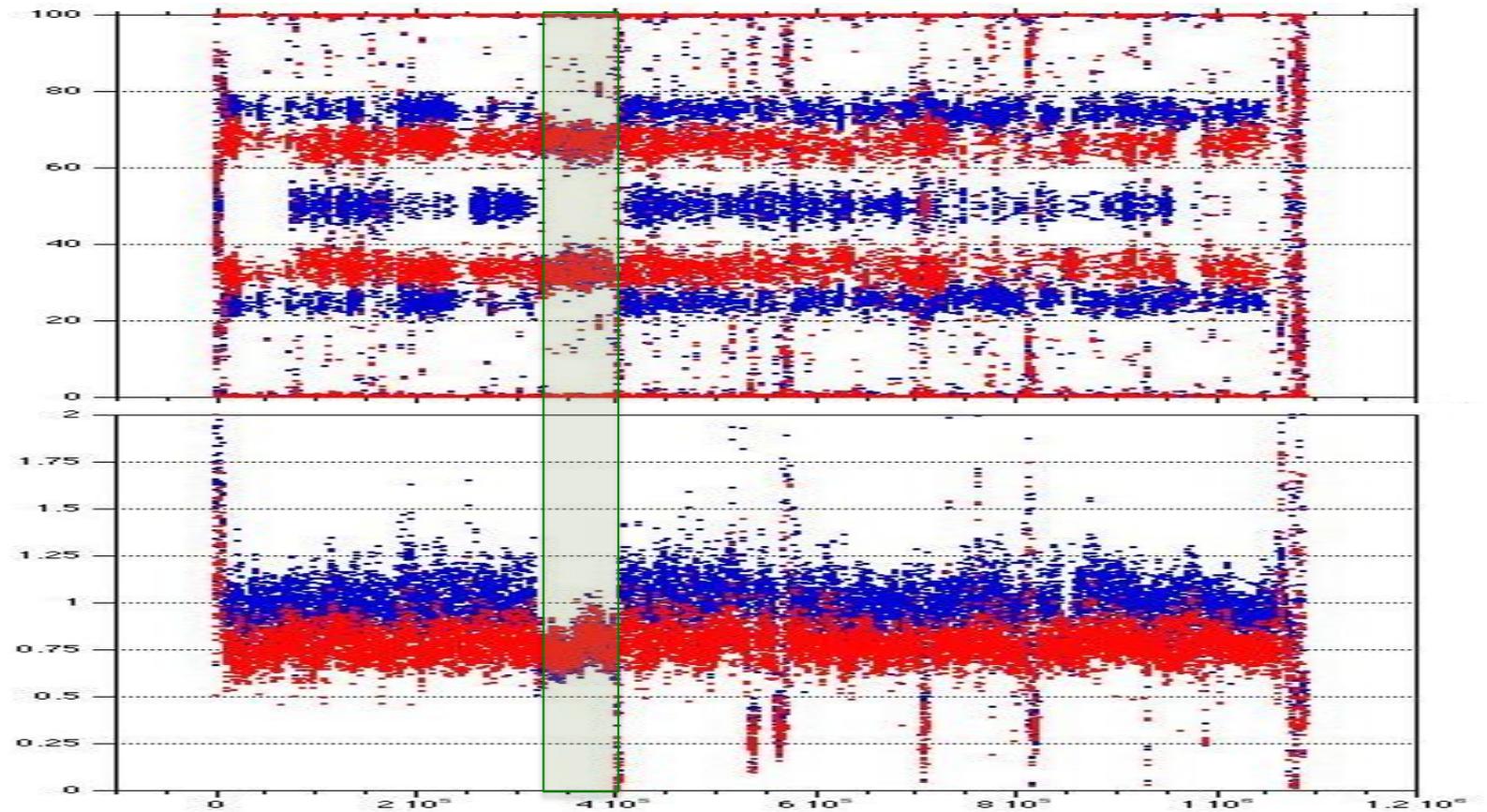
Chromosomes II, IX and XIII are tetraploid in both strains,
chromosome VIII is pentaploid in JBA-wt.

All other chromosomes are triploid in JBA-mut and tetraploid in JBA-wt.

Chromosome VII as a detailed example:

Upper plot shows frequency of each nucleotide at SNV-site. The wild type (**blue**) mainly has values being multiples of 25% indicative of a tetraploid. For the mutant (**red**) the values are multiples of 33%, in agreement with triploid.

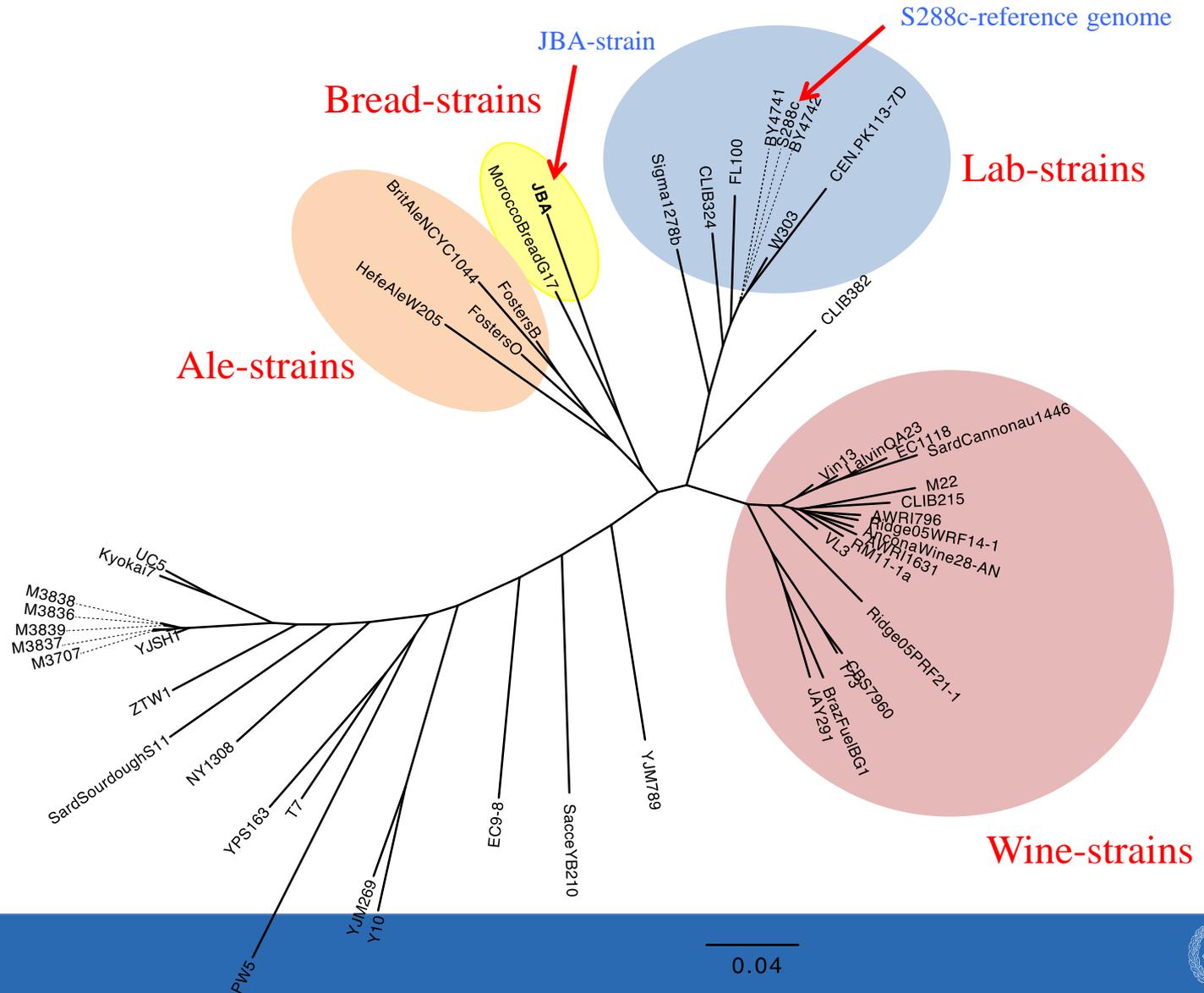
Lower plot shows normalized coverage values (**blue-wt**, **red-mut**), green square indicates a region of local triploidy in wt. Several regions of lower coverage, corresponding to regions lacking from one or more chromosome copies can be seen. (these are mostly transposable elements)



JBA is closely related to bread-baking strains

JBA-wt differs from JBA-mut by ~ 16000 heterozygous SNV's (1.3 SNV/1000 bp)

JBA-wt differs from the S288c reference genome by ~ 100 000 heterozygous SNV's (8.6 SNV/1000 bp)

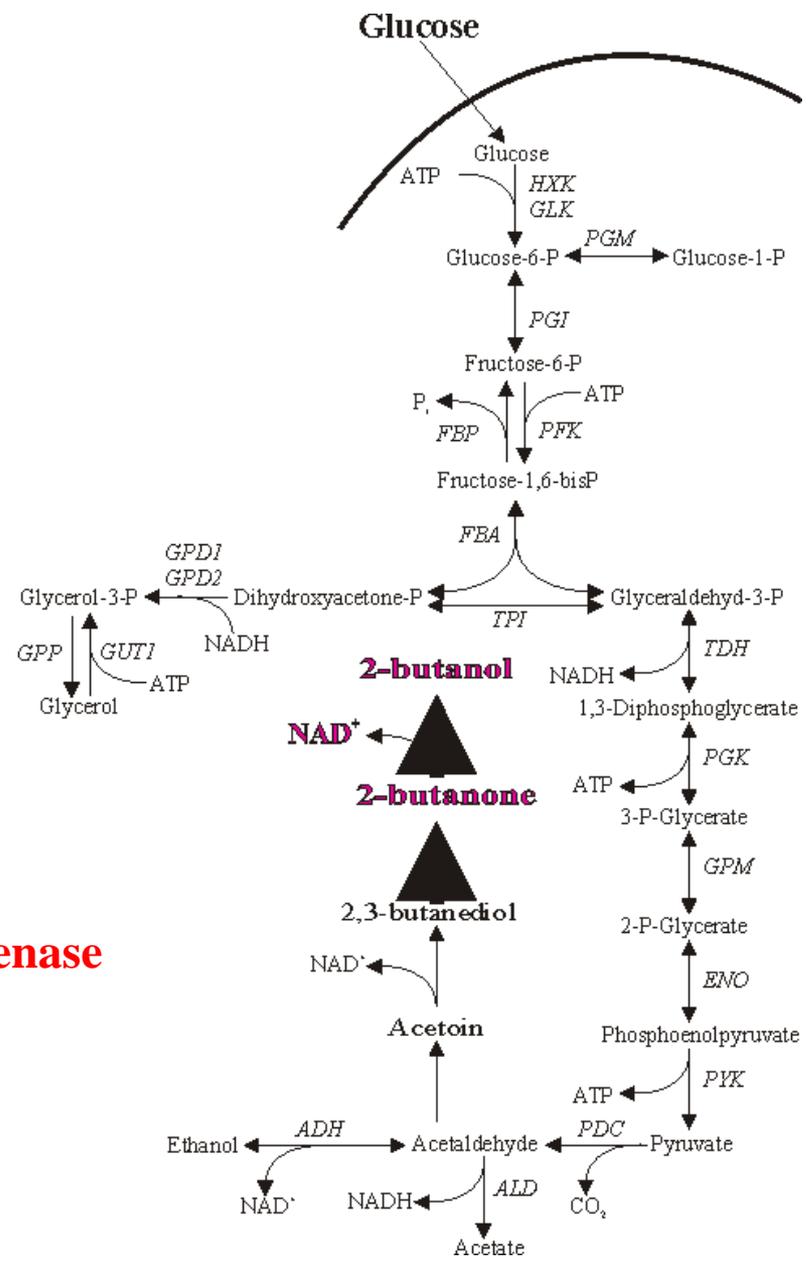
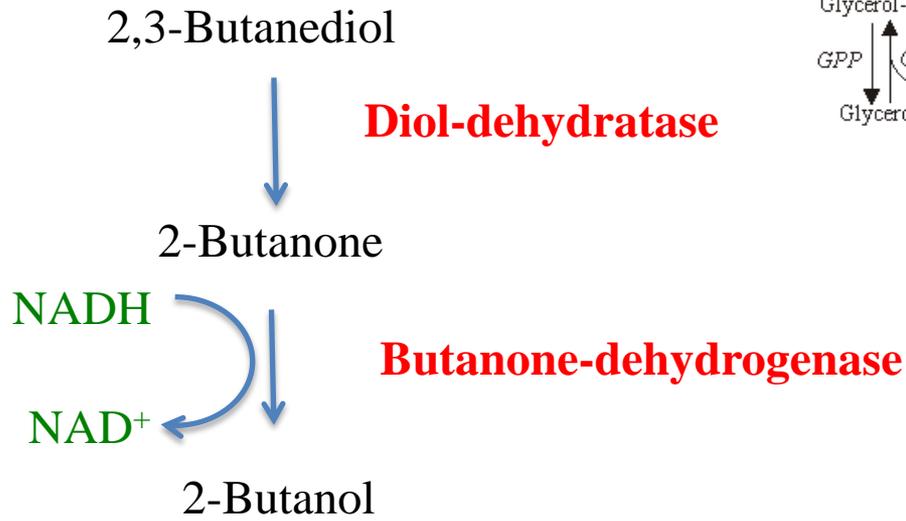


Conclusion from genome sequencing:

There are major changes in the genomic composition of JBA-wt and JBA-mutant which was surprising since there were few changes in the proteome

No differences in genes corresponding to proteins with an altered expression level between wild-type and mutant could be detected

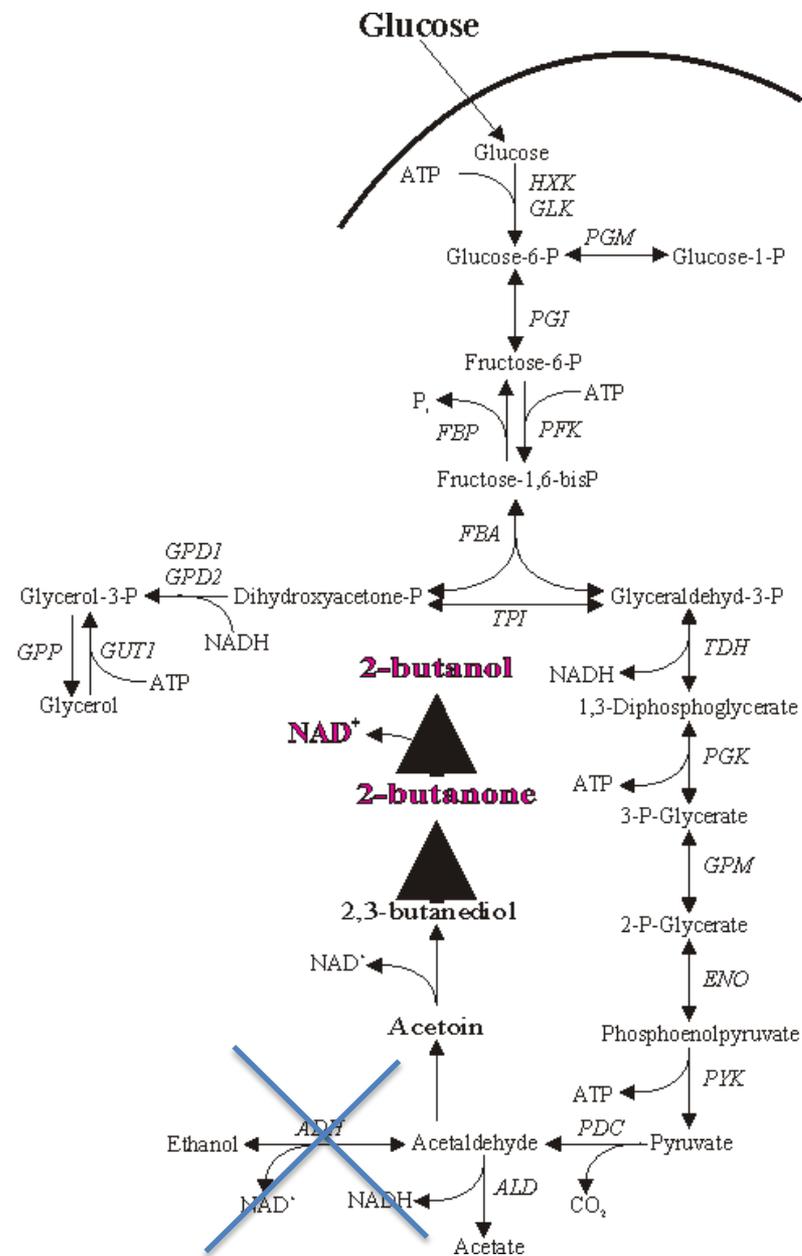
Part 2: 2-butanol production in yeast



Part 2: 2-butanol production in yeast

Butanol producing pathway that is very similar to the ethanol production pathway, e.g:

- Redox neutral
- Energetics is similar
- 2 ATP/Glucose
- Formation of one 4-carbon compound instead of two 2-carbon compounds



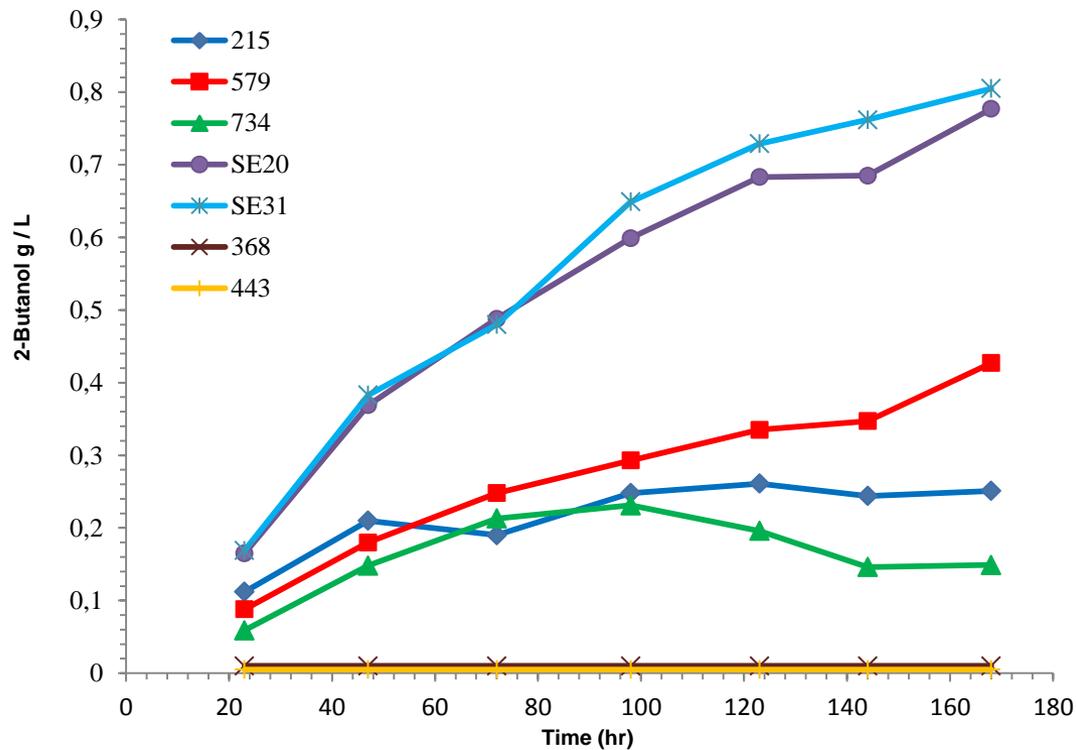
How to find the required enzymes?

The required enzyme activities are reported to be present in Lactobacilli and specifically *L. Brevis*

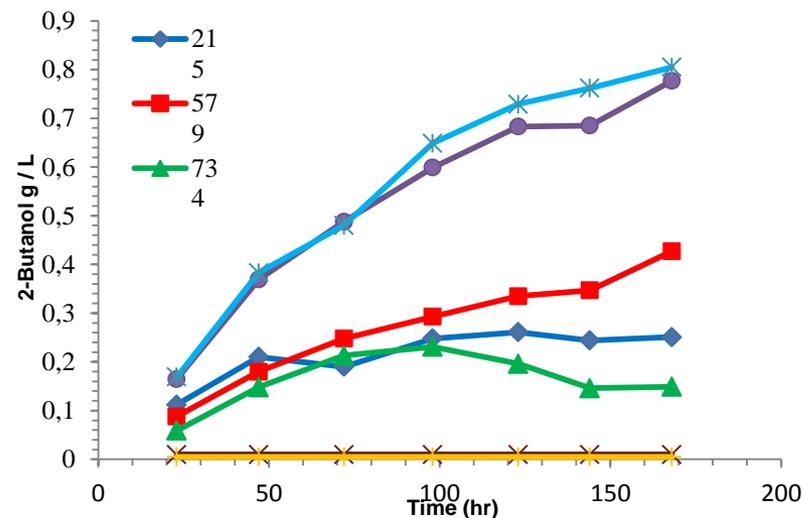
No conversion of 2,3-butanediol into 2-butanol in MRS medium but when using a defined medium (SM2).....

Strain		Source	<i>meso</i> -2,3-butanediol consumption	2-butanone consumption
<i>L. plantarum</i>	16 strains	ethanol pilot plant, Örnköldsvik	No	Not tested
<i>L. pantheris</i>	8 strains	ethanol pilot plant, Örnköldsvik	No	Not tested
<i>L. rossiae</i>	2 strains	ethanol pilot plant, Örnköldsvik	No	Not tested
<i>L. paracasei</i>	1 strain	ethanol pilot plant, Örnköldsvik	No	Not tested
<i>L. fermentum</i>	1 strain	ethanol pilot plant, Örnköldsvik	No	Not tested
<i>L. paracollinoids</i>	1 strain	ethanol pilot plant, Örnköldsvik	No	Not tested
<i>L. malefermentas</i>	1 strain	CCUG 32206	No	Not tested
<i>L. buchneri</i>	LB 12	ethanol pilot plant, Örnköldsvik	Yes	Yes
	LB 16	ethanol pilot plant, Örnköldsvik	Yes	Yes
<i>L. brevis</i>	SE 20	ethanol pilot plant, Örnköldsvik	Yes	Yes
	SE 31	ethanol pilot plant, Örnköldsvik	Yes	Yes
	LB 215	CCUG 21531	Yes	Yes
	LB 219	CCUG 21959	Yes	Yes
	LB 350	CCUG 35039	Yes	Yes
	LB 368	CCUG 36840	Yes	Yes
	LB 399	CCUG 39980	Yes	Yes
	LB 443	CCUG 44317	Yes	Yes
	LB 579	CCUG 57950	Yes	Yes
	LB 734	CNRZ 734	Yes	Yes

Production of 2-butanol from 2,3-butanediol by different isolates of *L. brevis*



Production of 2-butanol from 2,3-butanediol by different isolates of *L. brevis*



The diol-dehydratase is induced by the presence of propanediol as well as butanediol and shows activity also with glycerol

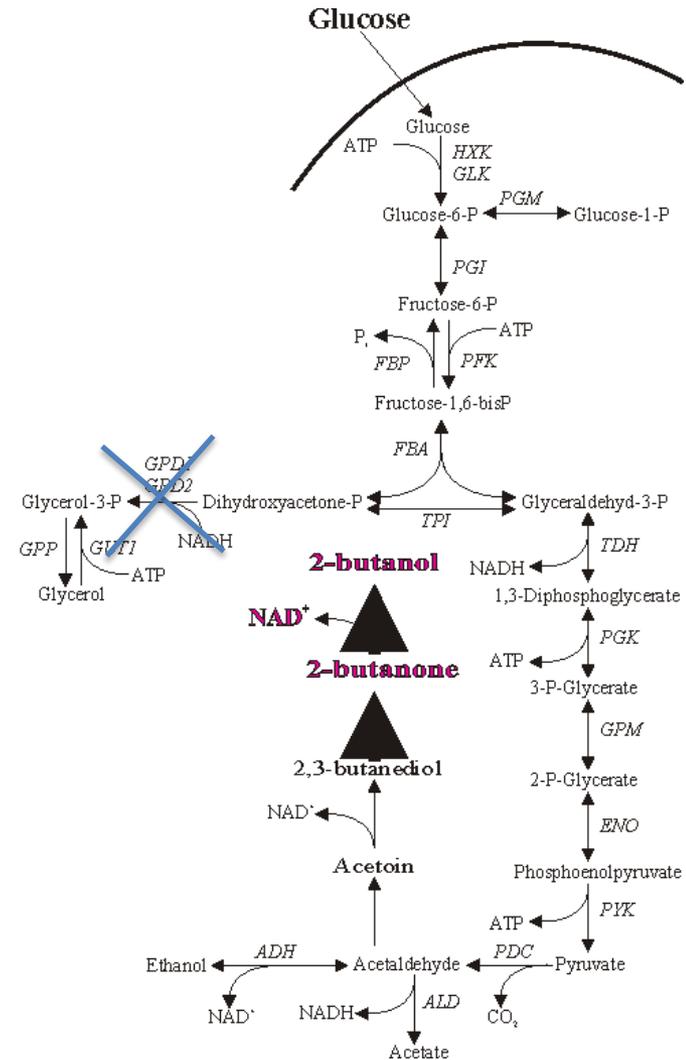
Grown in	Specific activity ($\mu\text{mol min}^{-1} \text{g}^{-1}$)			
	1,2-propanediol	2,3-butanediol	glycerol	1,3-propanediol
SM2	< 0.1	< 0.1	< 0.1	< 0.1
SM2 + 1,2-propanediol	0.19 ± 0.001	0.08 ± 0.001	0.10 ± 0.001	0.11 ± 0.001
SM2 + 2,3-butanediol	0.13 ± 0.001	0.10 ± 0.002	0.22 ± 0.001	0.16 ± 0.001

Introduction of diol-dehydratase and a secondary alcohol dehydrogenase into *S. cerevisiae*

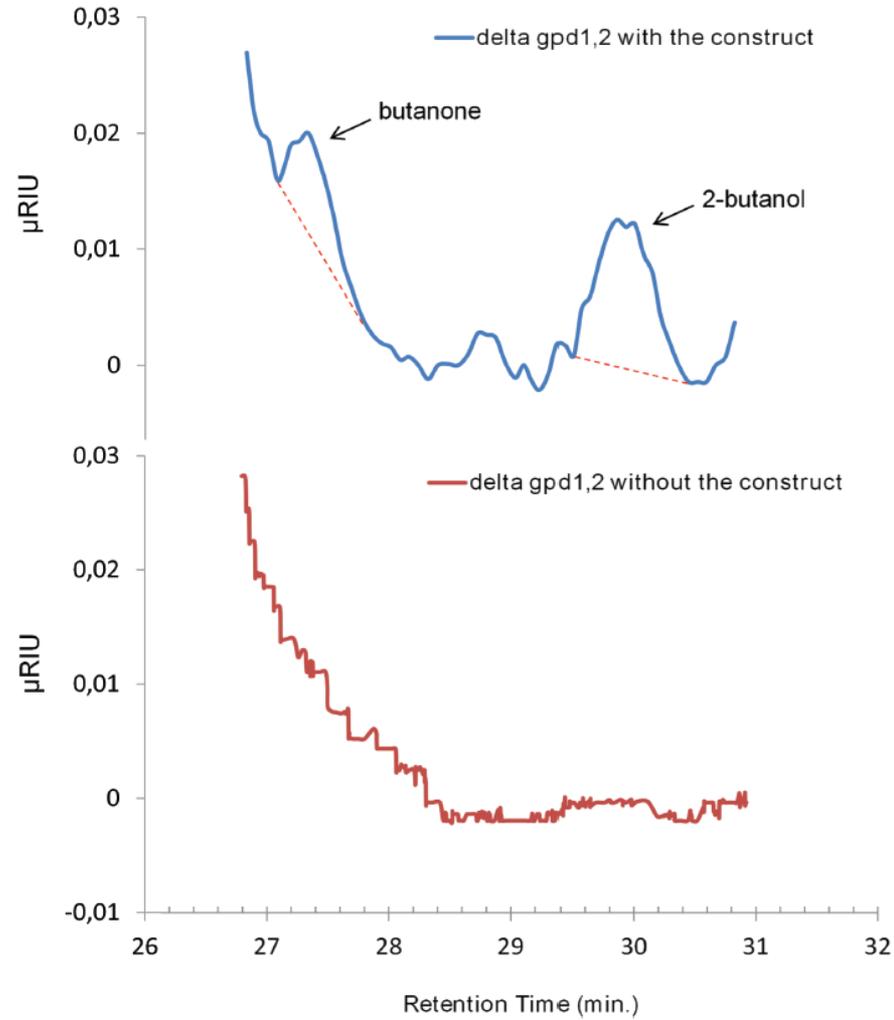
2-butanol production from 2,3-butanediol in a $\Delta gpd1,2$ double mutant

Wild type strain did not produce any 2-butanol

Diol-dehydratase is a vitamin B₁₂ dependent enzyme



2-butanol production from 2,3-butanediol in a $\Delta gpd1,2$ double mutant



Conclusions

- Evolutionary engineering increased the 2-butanol tolerance with more than 50%
- The evolved strain showed an enhanced tolerance to other isomers of butanol as well as other alcohols
- The evolved strain showed an enhanced expression of many mitochondrial and respiratory proteins
- Increased expression of glycerol-3-phosphate phosphatase (Gpp2) increased 2-butanol tolerance in yeast
- The capacity to produce 2-butanol from 2,3-butanediol seems widespread among strains of *L. Brevis*
- B₁₂ dependent 2-butanol production is established in *S. cerevisiae*

Transforming *Saccharomyces cerevisiae* into an ethylene producing organism

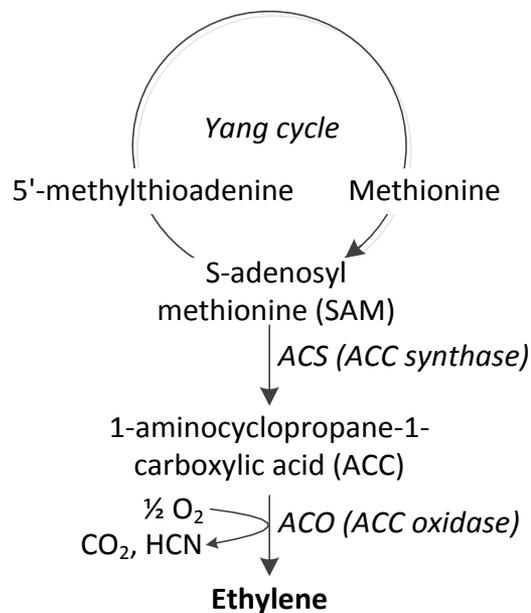
- **Nina Johansson**
- **Paul Quehl**
- **Karl Persson**
- **Joakim Norbeck**
- **Christer Larsson**
- *Chalmers University of Technology, Dept Chemical and Biological Engineering - Systems Biology, Gothenburg, Sweden*

Why ethylene in yeast?

- Bulk chemical (124 million tons/year)
- Fossil based production
- Yeast has an outstanding history of human usage and exploitation under large-scale industrial conditions
- Can cope with harsh and/or nutrient poor conditions such as, *e.g.* Lignocellulosic substrates
- Amenable to genetic manipulations

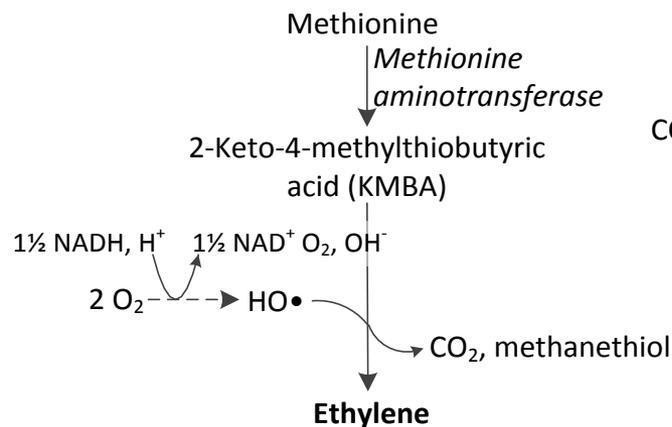
Three different ethylene producing pathways are identified

Plant pathway

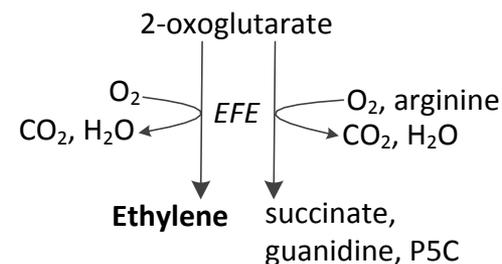


Microbial pathways

KMBA pathway



EFE pathway



EFE = Ethylene Forming Enzyme and the EFE pathway

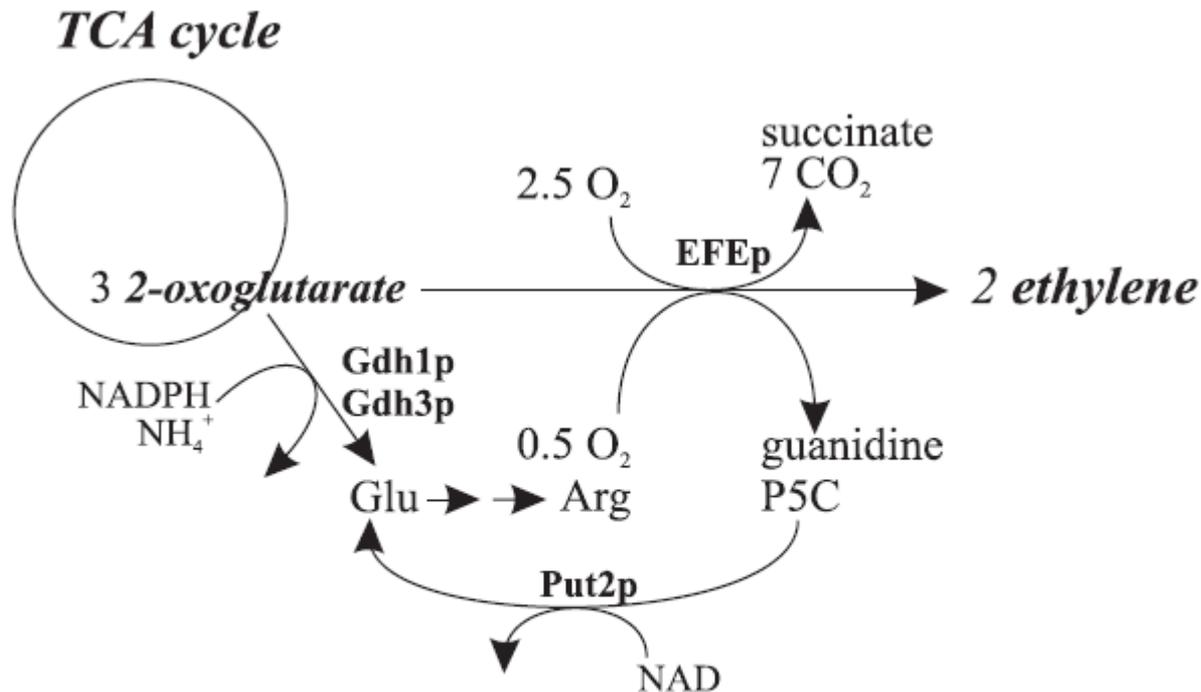
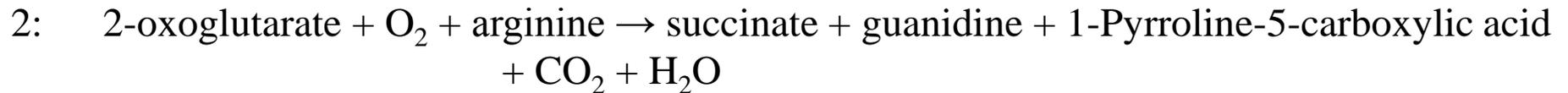
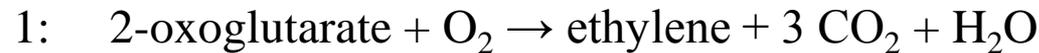
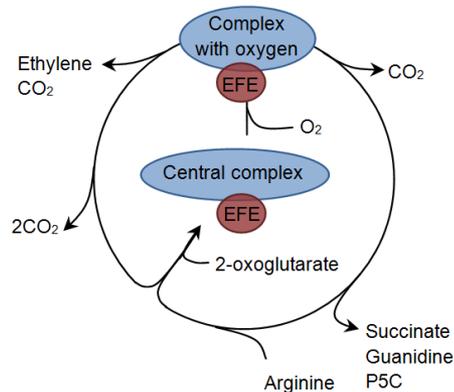


Fig. 1. Ethylene biosynthetic metabolism by insertion of the ethylene forming enzyme (EFE) from the plant pathogenic bacterium, *Pseudomonas syringae* (Fukuda et al., 1992) in the yeast *S. cerevisiae*.

Dual circuit mechanism of EFE



(Proposed by Fukuda et al. (1992) Biochem. Biophys. Res. Com. 188:483-489)

Metabolic modelling identified oxygen and respiration of NADH as key factors for efficient ethylene production

Ethylene production at different levels of dissolved oxygen. Chemostat cultivation at $D = 0.1\text{h}^{-1}$

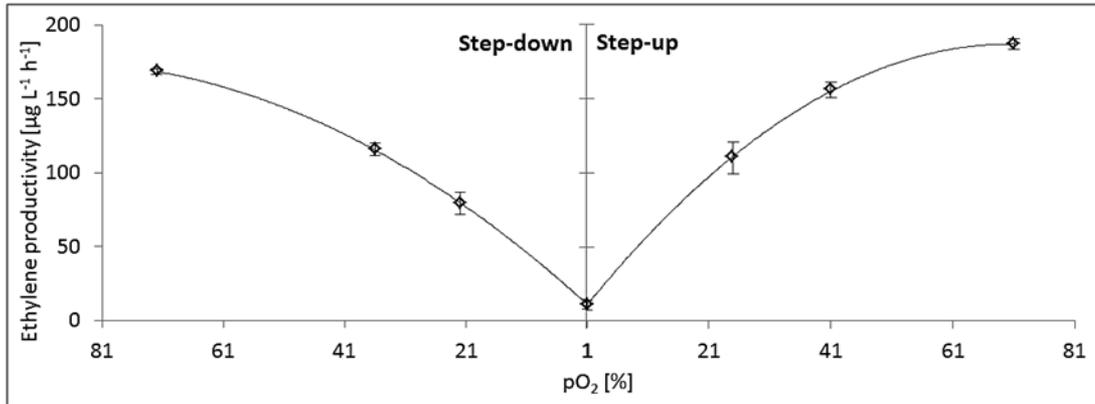


Table 2

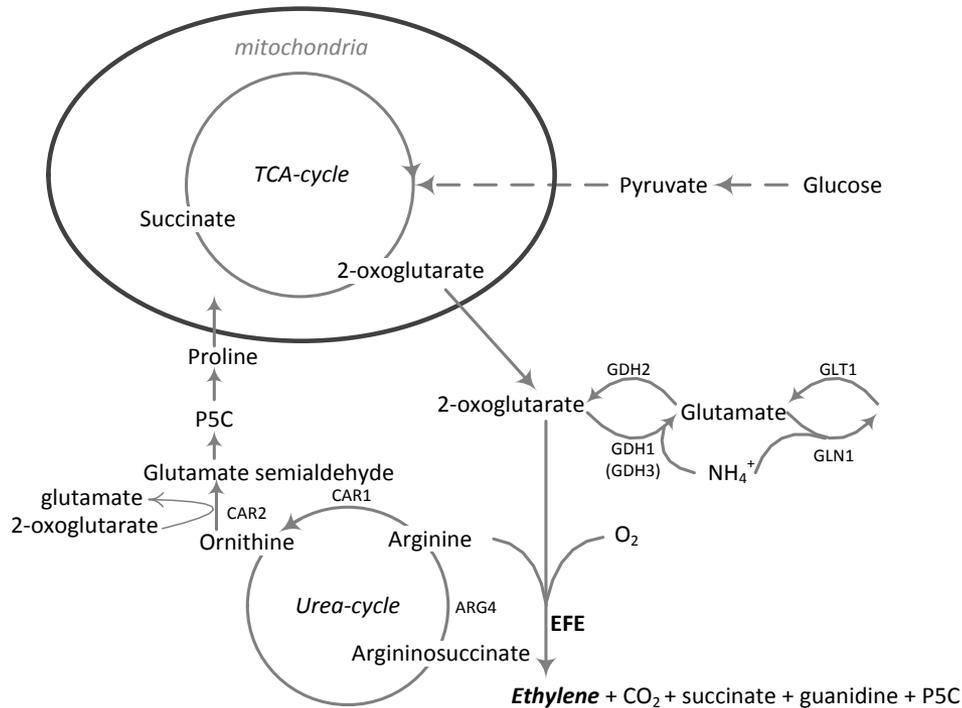
The effect of respiration rate on ethylene formation

Condition	Ethylene	
	Specific productivity [µg g _{DW} ⁻¹ h ⁻¹]	Productivity [µg L _{Culture} ⁻¹ h ⁻¹]
Reference condition	30.4 ± 2.8	178 ± 25
+ 7.5 mM Benzoate	50.3 ± 1.3	37.7 ± 0.9
+ 1 mM Azide	0	0

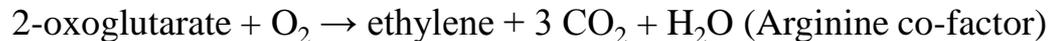
Oxygen important due to:

- 1. EFE poor affinity for O₂**
- 2. Required for NADH oxidation**

Close connection between ethylene formation and carbon as well as nitrogen metabolism



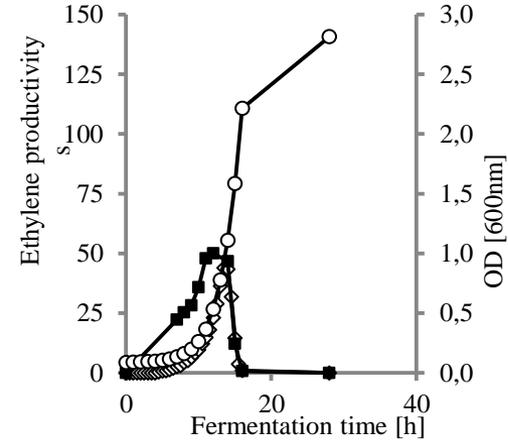
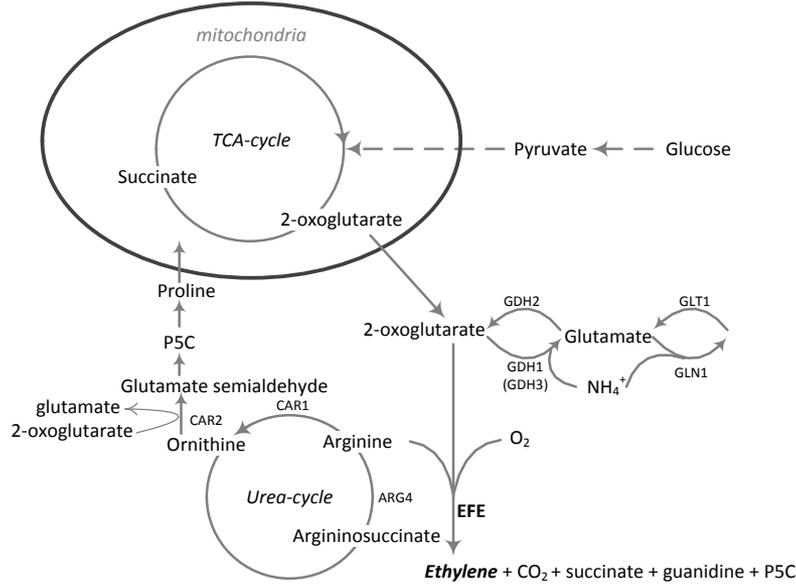
- Glutamate instead of NH_4 drastically improves productivity and yield of ethylene
- Arginine drastically reduces productivity and yield
- Absence of arginine blocks ethylene formation



Metabolic engineering strategies to enhance ethylene productivity and yield by decreasing intracellular levels of arginine

Overexpression of CAR1 \longrightarrow No effect

Deletion of ARG4

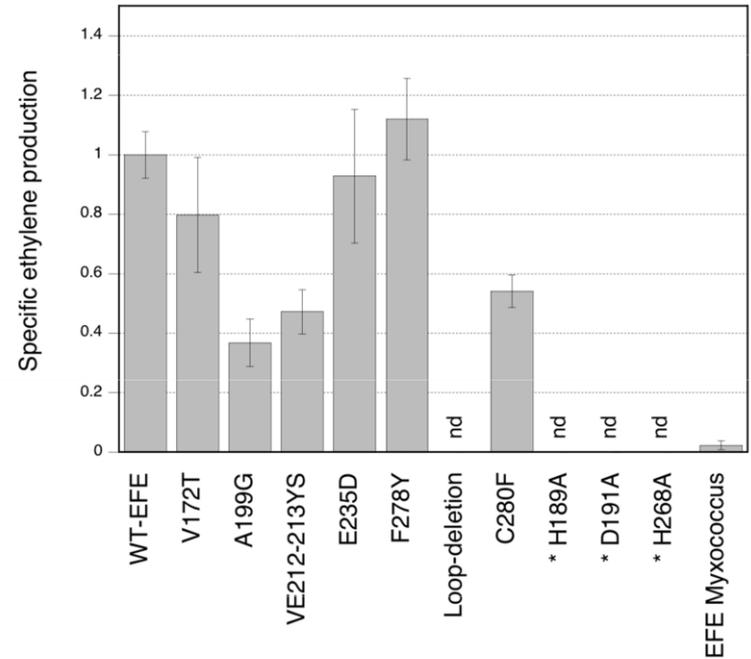
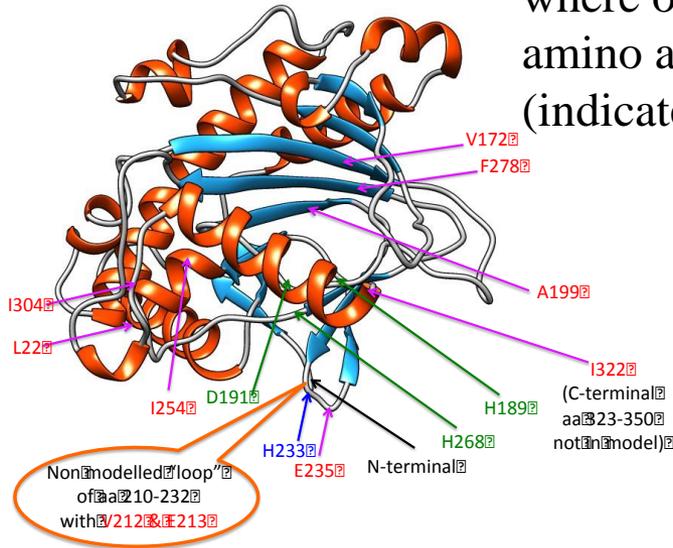


Arginine as a nitrogen source induced formation of pyruvate? Extracellular levels above 1 g/l

Structure function studies of EFE

Attempts to determine structure using NMR and crystallography failed!
But
Predicted structure from sequence homology with a dioxygenase from
Arabidopsis thaliana

Comparison of amino acid sequence of three EFE's
where one did not produce ethylene identified
amino acids correlating with ethylene production
(indicated in the figure)



Acknowledgements



Financial support from the Swedish Energy Agency, Chalmers Energy Area of Advance and the European community's seventh framework programme (FP7-241566-BIOCORE) is gratefully acknowledged

Thank you and questions?